

FORM PTO-1390
(REV. 9-2001)

U.S. DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE

ATTORNEY'S DOCKET NUMBER

01760/HG

TRANSMITTAL LETTER TO THE UNITED STATES
DESIGNATED/ELECTED OFFICE (DO/EO/US)
CONCERNING A FILING UNDER 35 U.S.C. 371

U.S. APPLICATION NO. (If known, see 37 CFR 1.5)

09/980413

INTERNATIONAL APPLICATION NO.
PCT/EP00/05471INTERNATIONAL FILING DATE
June 14, 2000PRIORITY DATE CLAIMED
June 18, 1999

TITLE OF INVENTION

LEATHER TANNING

APPLICANT(S) FOR DO/EO/US Christopher Raymond JONES, Gareth Rhys COLLINS,
and Robert Eric TALBOT

Applicant herewith submits to the United States Designated/Elected Office (DO/EO/US) the following items and other information:

1. ☒ This is a **FIRST** submission of items concerning a filing under 35 U.S.C. 371.
2. ☐ This is a **SECOND** or **SUBSEQUENT** submission of items concerning a filing under 35 U.S.C. 371.
3. ☐ This is an express request to begin national examination procedures (35 U.S.C. 371(f)). The submission must include items (5), (6), (9) and (21) indicated below.
4. ☒ The US has been elected by the expiration of 19 months from the priority date (Article 31).
5. ☒ A copy of the International Application as filed (35 U.S.C. 371(c)(2))
 - a. ☐ is attached hereto (required only if not communicated by the International Bureau).
 - b. ☒ has been communicated by the International Bureau.
 - c. ☐ is not required, as the application was filed in the United States Receiving Office (RO/US).
6. ☐ An English language translation of the International Application as filed (35 U.S.C. 371(c)(2)).
 - a. ☐ is attached hereto.
 - b. ☐ has been previously submitted under 35 U.S.C. 154(d)(4).
7. ☐ Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. 371(c)(3))
 - a. ☐ are attached hereto (required only if not communicated by the International Bureau).
 - b. ☐ have been communicated by the International Bureau.
 - c. ☐ have not been made; however, the time limit for making such amendments has NOT expired.
 - d. ☐ have not been made and will not be made.
8. ☐ An English language translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371 (c)(3)).
9. ☐ An oath or declaration of the inventor(s) (35 U.S.C. 371(c)(4)).
10. ☐ An English language translation of the annexes of the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371(c)(5)).

Items 11 to 20 below concern document(s) or information included:

11. ☒ An Information Disclosure Statement under 37 CFR 1.97 and 1.98, and substitute Form PTO-1449A
12. ☐ An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included.
13. ☒ A **FIRST** preliminary amendment.
14. ☐ A **SECOND** or **SUBSEQUENT** preliminary amendment.
15. ☐ A substitute specification.
16. ☒ A change of power of attorney and/or address letter.
17. ☐ A computer-readable form of the sequence listing in accordance with PCT Rule 13ter.2 and 35 U.S.C. 1.821 - 1.825.
18. ☐ A second copy of the published international application under 35 U.S.C. 154(d)(4).
19. ☐ A second copy of the English language translation of the international application under 35 U.S.C. 154(d)(4).
20. ☒ Other items or information:

- (i) WO 00/79011 A1 (Publication of Appln.)
- (ii) PCT/ISA/210 (Search Report)
- (iii) PCT/IB/308 (Notice to the Designated Offices)-Appln sent to U.S.
- (iv) PCT/RO/101 (PCT Request)
- (v) PCT/IPEA/416 (Notice of Transmittal of Preliminary Examination Report)
- (vi) PCT/IPEA/409 (International Preliminary Examination Report)
- (vii) PCT/IB/332 (Information re Elected Offices)
- (viii) PCT/IB/304 (Notice of Transmittal of Priority Document)
- (ix) REQUEST FOR PUBLICATION OF ASSIGNMENT INFORMATION

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No.: EL 867 738 761 US

Date of Deposit: November 30, 2001

I hereby certify that this paper is being deposited with the United States Postal Service Express Mail Post Office to Addressee's service under 37 CFR 1.10 on the date indicated above and is addressed to the U.S. Patent and Trademark Office, P.O. Box 2327, Arlington, VA 22202



Laraine Dobies

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21. ☐ The following fees are submitted:**BASIC NATIONAL FEE (37 CFR 1.492 (a) (1) - (5)):**

Neither international preliminary examination fee (37 CFR 1.482)
nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO
and International Search Report not prepared by the EPO or JPO..... \$1040.00

International preliminary examination fee (37 CFR 1.482) not paid to
USPTO but International Search Report prepared by the EPO or JPO\$890.00

International preliminary examination fee (37 CFR 1.482) not paid to USPTO
but international search fee (37 CFR 1.445(a)(2)) paid to USPTO \$740.00

International preliminary examination fee (37 CFR 1.482) paid to USPTO
but all claims did not satisfy provisions of PCT Article 33(1)-(4) \$710.00

International preliminary examination fee (37 CFR 1.482) paid to USPTO
and all claims satisfied provisions of PCT Article 33(1)-(4) \$100.00

ENTER APPROPRIATE BASIC FEE AMOUNT =

CALCULATIONS PTO USE ONLY

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Surcharge of \$130.00 for furnishing the oath or declaration later than ☐ 20 ☐ 30
months from the earliest claimed priority date (37 CFR 1.492(e)).

\$ --

CLAIMS	NUMBER FILED	NUMBER EXTRA	RATE	\$
Total claims	7 - 20 =	0	x \$18.00	\$ --
Independent claims	1 - 3 =	0	x \$84.00	\$ --
MULTIPLE DEPENDENT CLAIM(S) (if applicable)			+ \$280.00	\$ --

TOTAL OF ABOVE CALCULATIONS =

\$ 890.00

☒ Applicant claims small entity status. See 37 CFR 1.27. The fees indicated above
are reduced by 1/2.

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SUBTOTAL =

\$ 890.00

Processing fee of \$130.00 for furnishing the English translation later than ☐ 20 ☐ 30
months from the earliest claimed priority date (37 CFR 1.492(f)).

\$ --

TOTAL NATIONAL FEE =

\$ 890.00

Fee for recording the enclosed assignment (37 CFR 1.21(h)). The assignment must be
accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31). \$40.00 per property +

\$ --

TOTAL FEES ENCLOSED =

\$ 890.00

Amount to be
refunded:

\$

charged:

\$

- a. ☒ A check in the amount of \$ 890.00 to cover the above fees is enclosed.
- b. ☐ Please charge my Deposit Account No. _____ in the amount of \$ _____ to cover the above fees.
A duplicate copy of this sheet is enclosed.
- c. ☒ The Commissioner is hereby authorized to charge any additional fees which may be required, or credit any
overpayment to Deposit Account No 06-1378. A duplicate copy of this sheet is enclosed.
- d. ☐ Fees are to be charged to a credit card. **WARNING:** Information on this form may become public. Credit card
information should not be included on this form. Provide credit card information and authorization on PTO-2038.

NOTE: Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to revive (37 CFR
1.137 (a) or (b)) must be filed and granted to restore the application to pending status.

SEND ALL CORRESPONDENCE TO:

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01933

PATENT TRADEMARK OFFICE

Date: November 30, 2001

SIGNATURE

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26,853

REGISTRATION NUMBER

09/980413
JC10 Rec'd P&T/PTO 3 0 NOV 2001

Attorney Docket No. 01760/HG

**IN THE UNITED STATES PATENT
AND TRADEMARK OFFICE**

Applicant(s): Christopher R. JONES et al.

Serial No. : To be assigned (U.S.
National Phase of
PCT/EP00/05471
filed June 14, 2000)

Filed : CONCOMITANTLY HEREWITH

For : LEATHER TANNING

Art Unit :

Examiner :

Express Mail Mailing Label
No.: EL 867 738 761 US
Date of Deposit: November 30, 2001

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Laraine Dobies

In the event that this Paper is late filed, and the necessary petition for extension of time is not filed concurrently herewith, please consider this as a Petition for the requisite extension of time, and to the extent not tendered by check attached hereto, authorization to charge the extension fee, or any other fee required in connection with this Paper to Account No. 06-1378.

ATTENTION BOX PCT

**PRELIMINARY AMENDMENT FILED CONCOMITANT
WITH NATIONAL PHASE PCT APPLICATION**

Assistant Commissioner for Patents
Washington, D.C. 20231

S I R :

This is a PRELIMINARY AMENDMENT filed in the above-referenced national phase PCT application.

Please amend the application as follows:

IN THE SPECIFICATION:

Please insert the following paragraph before the first paragraph of the specification:

--This application is a U.S. National Phase Application under 35 USC 371 of International Application PCT/EP00/05471 (published in English) filed June 14, 2000.--

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IN THE CLAIMS:

Please cancel claims 1-7 (all of the claims in the case) and add new claims as follows:

8. (New) A mixture for use in tanning leather, said mixture comprising:

(I) a phosphorus-containing compound selected from the group consisting of

tris(hydroxymethyl)phosphine (THP);

tetrakis(hydroxymethyl)phosphonium salts (THP⁺salts);

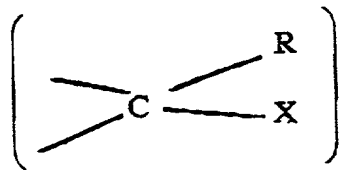
condensates of THP with organic nitrogen compounds (THP condensates); and

compounds of the formula $RP(CH_2OH)_3$, wherein R is an organic group which does not react chemically with collagen in said leather (THP analogues); and

(II) at least one water-soluble moderator selected from the group consisting of

(A) metaphosphate salts and

(B) aliphatic polyhydroxy or polypyrrolidone compounds having at least three adjacent, geminal or vicinal



groups,

wherein R is selected from the group consisting of hydrogen and alkyl and wherein X is selected from the group consisting of hydroxyl and pyrrolidone.

9. (New) The mixture of claim 8, wherein said phosphorous containing compound consists essentially of a condensate of from 2 to 6 moles THP with 1 mole urea.

10. (New) The mixture of claim 8, wherein said at least one water-soluble moderator is selected from the group consisting of sorbitol, glycerol, trimethylolpropane, pentaerythritol, mannitol, mono and di- saccharide sugars, dialdehyde starch, alginates, polyvinyl alcohol and polyvinyl pyrrolidone.

11. (New) A method of tanning leather which comprises contacting the collagen in said leather with a tannage comprising the mixture of claim 8.

12. (New) The method of claim 11, further comprising tanning said leather with a syntan and/or a mineral tannage and wherein said mixture is contacted with the collagen in said leather prior to, simultaneously with, or after tanning of said leather with the syntan and/or mineral tannage.

13. (New) The method of claim 11, wherein said leather is fat liquored subsequent to said tanning.

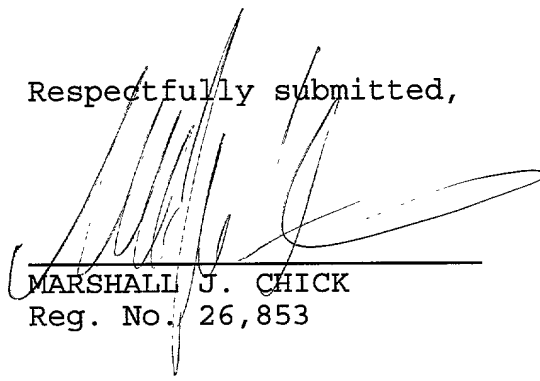
14. (New) The method of claim 11, wherein said leather is shaved and/or split subsequent to said tanning.

R E M A R K S

Entry of this AMENDMENT and a favorable action on the merits
are respectfully requested.

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Respectfully submitted,



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LEATHER TANNING

The present invention relates to a novel process for tanning leather.

Tanning is a process for the preservation of skins, by which is meant the collagen-containing integuments of vertebrates including mammals (e.g. cows, pigs, deer, goats, sheep, seals, antelope, mink, stoats and camels), fish, (e.g. sharks), reptiles (e.g. snakes, lizards and crocodiles), and birds (e.g. ostrich). Skins comprise a layer of collagen, and tanning entails reacting the collagen with a cross linking, or tanning, agent to cross link reactive sites within the collagen molecule. The product of the cross linking is leather, which is substantially less susceptible than skin to bacterial degradation.

A consequence of the cross linking is an increase in the minimum temperature at which the wet leather tends to shrink. This shrink temperature is often used as an indication of the degree of tanning.

The collagen layer of the skin is typically separated from fats, connective tissue and other subcutaneous protein, and optionally from the outer keratinous layer, by a combination of chemical and physical steps. The former may include liming, bating, pickling and/or degreasing.

The skin is then subjected to treatment in one or more stages with various tanning agents selected to give the desired end properties.

The main types of tannage are:- vegetable tannage, based on tannin as the active cross linking agent; mineral tannage using various polyvalent metal salts, especially salts of chromium, aluminium, iron, or zirconium; and synthetic tanning agents, referred to as "syntans". Syntans include replacement syntans which are active tanning agents capable of tanning leather when used as the sole tannage e.g. by reacting with collagen at two or more sites to form cross links, and auxiliary syntans

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which are added to other tannages to modify the character of the leather but which are not in themselves active tanning agents. Auxiliary syntans are absorbed by the leather or may react with collagen at one site only. Syntans include various polymers and copolymers, such as those obtained by condensing formaldehyde with, for example, phenols and/or aryl sulphonates, and acrylate, methacrylate, acrylamide and/or acrylonitrile homopolymers and copolymers. Formaldehyde itself and dialdehydes such as glutaraldehyde are also used in tanning, usually in combination with other tannages.

For centuries the production of leather was based on the vegetable tannages which produce the characteristic brown colour traditionally associated with leather. One of the first mineral tannages was alum, but currently the most widely used tanning agent is chrome, usually in the form of basic chromium sulphate, which produces a blue grey leather with high shrink temperatures. However mineral tannages in general, and chrome tannages in particular are under pressure on environmental grounds. Syntans are less environmentally harmful than mineral tanning agents.

Tetrakis (hydroxymethyl) phosphonium salts which will be referred to herein as "THP salts" have long been used as fire-retardants for textiles and have been applied to the keratinous (fur) side of skins for this purpose. The salts may be applied directly to the fabric or in the form of precondensates which are water soluble or sparingly water soluble copolymers of THP with organic nitrogen compounds such as urea or an amine and which are referred to herein as "THP condensates". THP salts have also been known as possible ingredients of tanning liquors for more than thirty six years. US 2 992 879 referred to THP chloride (THPC) as an unsatisfactory tanning agent on its own, and recommended a combination of THPC and a phenol such as resorcinol speculating that the two react together to form an effective tanning agent when the pH is raised. In fact THP salts do copolymerise with phenols such as resorcinol (see, for example, Textile Research Journal, December 1982, P743). US 3 104 151 describes the use of such THPC phenol copolymers as pretannages for leather in which the main tannage is vegetable or mineral. GB 2 287 953 describes the use of THP salts as cross linkers in conjunction with melamine formaldehyde or

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urea formaldehyde prepolymers, in order to form a copolymeric tanning agent in situ in the tanning liquor. EP 0 559 867 describes the use of phosphonium salts such as THP sulphate (THPS) on raw or cured skin prior to tanning e.g. in acid degreasing. EP 0 681 030 describes the use of THPS as a cross linker for casein finishes applied to leather after tanning. GB 2 314 342 describes the use of hydroxyalkyl phosphines and phosphonium salts as tanning agents in conjunction with aromatic anionic syntans and EP 0 808 908 describes the use of THP salts with condensable nitrogen compounds.

THP salts are stable under acidic conditions in the absence of air or oxidising agents. At pH above 3 and in the absence of oxidising agents they are gradually converted to the parent base, tris(hydroxymethyl)phosphine commonly referred to as THP. Conversion is rapid and substantially complete between pH of about 4 and 6. Above pH 7, or in the presence of oxidising agents THP salts or THP are converted to tris(hydroxymethyl)phosphine oxide (THPO), conversion being rapid and substantially complete at pH above about 10, e.g. 12. It has been stated, e.g. in US 2 993 744, that THPO is the effective tanning agent in THP based tannages.

Contrary to statements in the art, THPO is not effective as a tanning agent for leather, and THP salts are also ineffective as tannages. Moreover THP used in conjunction with co-condensable monomers or polymers provides complex systems which are difficult to control to obtain consistent results.

WO99/23261 describes the use of THP on its own as an effective main tanning agent. THP is usually most effective when formed in situ by first impregnating the leather with a THP salt, in the substantial absence of monomers or prepolymers which react or copolymerise with THP and raising the pH above 4 and preferably above 5. WO99/23261 also describes the use of THP condensates as main tanning agents.

In addition to THP, compounds of the formula $R P(CH_2OH)_2$, referred to herein as "THP analogues", where R is an organic group which does not react chemically with collagen, such as a C_{1-20} alkyl, alkenyl, aryl, aralkyl, alkaryl, polyalkyleneoxy,

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alkylpolyalkyleneoxy or polyalkyleneoxy alkyl group have been referred to in the literature as highly effective tanning agents.

For convenience "THP" will be used herein, where the context permits, to refer generically to THP, THP salts, THP condensates and THP analogues.

A major problem when tanning with derivatives of formaldehyde, including formaldehyde condensate syntans and THP is the evolution of formaldehyde during tanning and the presence of formaldehyde residues in the leather which give rise to objectionable odours and may cause a health hazard to process operators.

The steps required to produce leather including the pickling and degreasing which usually precede tanning, remove most of the natural oils and fats from leather. These are normally at least partially replaced after tanning by fat liquoring, which entails contacting the leather with an aqueous emulsion of oils and fats which soften and lubricate the finished leather. It has been found that leather which has been tanned with THP alone is tight and tends to be relatively hard. It requires high levels of fat liquor to soften it. It would be useful to be able to obtain the advantages of THP, but prepare a fuller, softer and more easily fat liquored leather.

It is often necessary to split, e.g. bovine leather, prior to retanning, in order to obtain the desired thickness. Conventionally tanned leather may split unevenly and the split leather may require substantial shaving to obtain a consistent thickness.

A further problem is swelling of the skin, control of which usually requires the presence of salts such as chloride or sulphate.

One object of the present invention is to provide tannages which exhibit the advantages of THP but provide leather which is fuller, softer and requires less fat liquor than leather tanned with THP alone.

A further object is to provide leather with lower formaldehyde levels during processing and on the finished leather, than are common in tannages based on

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formaldehyde derivatives. Another object is to provide leather which gives a more even split than conventionally tanned leather.

A further object is to obtain leather with reduced tendency to swell, compared with normal THP tannage.

We have now discovered that mixtures of THP with certain moderators, which appear to interact with THP to mask its OH groups at least partially, can be applied to skins to provide full soft leather which is readily fat liquored, contains reduced formaldehyde levels, is more easily split and shaved and exhibits reduced swelling tendencies.

We have found for example that a mixture of THP with a water soluble metaphosphate or with a water soluble polyhydroxy or polypyrolidone compound containing at least three adjacent, geminal or vicinal $>CR OH$ groups or pyrolidone groups provides improved tanning compared with THP alone, and in particular provides a soft, less tight and more easily fat liquored product.

The leather also shows reduced tendency to swelling which may be controlled without the use of salts.

In addition formaldehyde levels, in the process and in the finished leather are substantially less using the aforesaid mixtures than are observed when THP is used alone.

It has also been found that leather tanned using the mixture is more easily shaved and split.

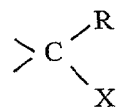
We believe that THP interacts chemically with the polyhydroxy compounds possibly by hydrogen bonding, and that this moderates the reaction with collagen.

Our invention therefore provides a mixture of THP as herein defined with at least one

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water soluble moderator selected from (A) metaphosphate salts and (B) aliphatic polyhydroxy or polypyrolidone compounds comprising at least three adjacent, geminal or vicinal



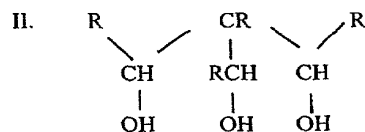
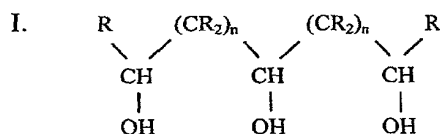
groups where R is hydrogen or an alkyl group, and X is a hydroxyl or

pyrolidone group.

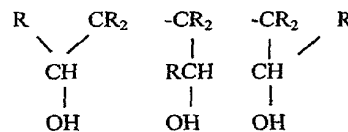
Our invention also provides a method of tanning or degreasing leather which comprises contacting collagen with a mixture as aforesaid.

Preferably the water soluble polyhydroxy or polypyrolidone compound has a molecular weight less than 200,000, more preferably less than 150,000, most preferably less than 100,000, e.g. less than 75,000. We especially prefer compounds with a mole weight less than 50,000.

The polyhydroxy compounds preferably comprise hydroxy methylene groups which may be adjacent (i.e. directly linked as in glycerol), geminal (i.e. linked to the same carbon atom as in pentaerythritol) or vicinal (i.e. linked to adjacent carbon atoms, as in trimethylolpropane). Thus the polyhydroxy compounds useful in the invention comprise compounds of the formulae:



and III.



wherein each n may independently be 0, 1 or 2 and each R is independently hydrogen

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or an alkyl, a hydroxy- or polyhydroxy- alkyl, a carbonyl, a carboxyl, a C₁₋₂₅ acyl, a carbonyl- carboxy- or C₁₋₂₅ acyl- substituted alkyl group or an alkyl group or hydroxy-, carboxy- carbonyl- or C₁₋₂₅ acyl substituted alkyl group which is interrupted by one or more ether or carbonyl linkages or wherein two or more R groups may together constitute one or more alicyclic rings, optionally substituted with one or more hydroxy, carbonyl, carboxy and/or acyl groups and/or interrupted by one or more ether or carbonyl linkages.

The aliphatic polyhydroxy compound may for example be glycerol, sorbitol, trimethylol propane, pentaerythritol, 1,2,3-trihydroxy butane, 1,2,3,4-tetrahydroxy butane, 1,2,3-trihydroxy pentane, 2,3,4-trihydroxy pentane, tetrahydroxy pentane, pentahydroxy pentane, water soluble carbohydrates including mono and disaccharides and soluble starches and alginates. Examples of suitable carbohydrates include sucrose, glucose, maltose, lactose, dextrose, fructose, xylose, arabinose, mannose, ribose and rhamnose and dialdehyde starch. The compound may also be a derivative of a carbohydrate including alkyl glycosides such as a C₁₋₂₅ alkyl glucoside or polyglucoside, or ascorbic, mannonic or gluconic acids, or alginates or sorbitol or mannitol or C₁₋₂₅ acyl esters of either sucrose or sorbitan. Generally we prefer not to use nitrogen containing derivatives.

A preferred group of polyhydroxy compounds comprises polyvinyl alcohols and, in particular, relatively low molecular weight polyvinyl alcohols. We prefer PVAs with a mean molecular weight less than 70,000, especially less than 50,000, e.g. less than 30,000.

Also highly effective in modifying the action of THP tannages is polyvinyl pyrrolidone which appears to function in an analogous manner to polyvinyl alcohol.

THP condensates containing high proportions of nitrogen and correspondingly low proportions of uncondensed THP, e.g. less than 4:1 molar THP:nitrogen compound are generally less effective with polyhydroxy or polypyrrolidone compounds than the uncondensed THP. However THP condensed with lower proportions of nitrogen

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compound e.g. condensates with greater than 4:1, especially greater than 5:1, for instance 5:1 to 7:1 molar THP/nitrogen compound, which contain some free THP, give particularly good results.

The metaphosphate salt is a cyclic condensed phosphate having the formula $(MPO_3)_n$ where M is preferably alkali metal or ammonia, e.g. potassium or most preferably sodium and n is from 2 to 10, preferably 3 to 8, e.g. 5 to 7. Particularly preferred is sodium hexametaphosphate. In principle any water soluble metaphosphate salt of a cation which does not react with THP may be used.

The relative proportion of THP and moderator may be equimolar or may comprise an excess of either component. Preferably the THP is in an excess of up to 20:1 molar, depending on the characteristics required. Generally the higher the proportion of polyhydroxy compound the less tight the leather. Proportions between 10:1 and 2:1 molar are generally preferred.

The THP and moderator may be supplied as or part of a formulated product, premixed or mixed in situ in the tanning operation.

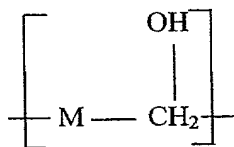
The THP may be used in conjunction with syntans, e.g. by treating the skin with said THP and a syntan (preferably a syntan which does not react with THP under normal tanning conditions) and/or by treating the skin in a plurality of stages at least one of which entails treatment with syntan and at least one other of which entails treatment with the THP.

For the purpose of this specification "syntan" is used to refer to replacement syntans which are synthetic organic compounds capable of reacting with collagen at two or more sites to form cross links and also to auxiliary syntans which do not in themselves contribute substantially to the cross linking but which are physically absorbed by the leather or react at no more than one site so as to modify the physical properties of the leather. For example the term includes any water soluble polymer

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prepared by copolymerising formaldehyde, which is capable of increasing the shrink resistance of collagen and which comprises at least two units of the formula



where each M is an aryl group such as a phenyl, naphthyl or aniline group substituted with one or more hydroxyl and/or sulphate, sulphone or sulphonimide groups or a urea or melamine residue. As used herein the term "syntan" also includes resin syntans which are homopolymers and copolymers of unsaturated carboxylic acids or their salts, esters, amides or nitriles, e.g. acrylic acid, methacrylic acid, acrylamide, acrylonitrile, maleic acid, fumaric acid, itaconic acid, aconitic acid, crotonic acid, isocrotonic acid, citraconic acid, mesaconic acid, angelic acid, tiglic acid and cinnamic acid. The copolymers may also comprise other vinylic comonomers such as styrene. Also included are acetone condensates with, for example sulphones and sulphonamides. Resin syntan may modify such properties as dye levelling, filling, grain appearance, break, grain strength, buffing nap, softness and tensile or tear strength.

The THP tends to react with nitrogen containing syntans and with other nitrogenous compounds and also to some extent with aromatic syntans. Such reactions may compete with cross linking reactions of both the THP and the syntan resulting in chaotic behaviour when the two are used together. Such systems are so highly sensitive to the conditions of use that it may be difficult to control them so as to obtain the consistent results required for a viable commercial process.

For these reasons we prefer not to use mixtures with such compounds. However it is possible to precondense the THP with, for example, cocondensable nitrogenous compounds such as urea, melamine, dicyandiamide and/or aliphatic amines to form condensates comprising two or more phosphorus atoms and having at least two hydroxymethyl groups.

High pH is preferably avoided because it converts the phosphine group to phosphine

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oxide, which is substantially ineffective as a tanning agent. We prefer that solutions for use according to our invention should have a pH less than 10, more preferably less than 9, especially less than 8, typically less than 7.5, most preferably 4.5 to 7. We prefer that the THP contains less than 15%, more preferably less than 10%, e.g. less than 9% of THPO based on the weight thereof. Generally the less THPO present the better.

The THP is preferably used in a total concentration of from 0.01 to 20% by weight based on the total weight of the tanning liquor, more preferably 0.5 to 10%, e.g. 1 to 5%, most preferably 1.5 to 4%. The total proportion of the THP used is preferably from 0.3 to 20% by weight based on the weight of wet skin, more preferably 1 to 15%, especially 1.5 to 10%, most preferably 2 to 5%.

We particularly prefer processes in which skins are treated with an auxiliary syntan prior to tanning with the THP.

When the mixture is used in combination with a syntan, the latter is preferably a polyacrylate, polymethacrylate, or copolymer of acrylic and/or methacrylic acid with acrylonitrile and/or acrylamide. Typically the polymer has a molecular weight in the range 1,000 to 200,000, more usually 3,000 to 100,000.

The syntan is preferably present at a concentration of from 0.5 to 35% by weight of the tanning liquor. e.g. 1 to 20%, more preferably 2 to 10% especially 3 to 6%. The total proportion of syntan used is preferably from 1 to 20% by weight based on the wet weight of skins, e.g. 2 to 10% especially 3 to 5%.

The proportion by weight of THP to syntan may typically be from 1:10 to 10:1, preferably 1:5 to 2:1, especially 1:2 to 1:1. The total proportion of tannages used is preferably from 2 to 20% active weight based on the wet weight of skins, e.g. 3 to 10%, especially 4 to 8%. The total tannage used preferably comprises more than 80% by weight, more preferably more than 90% by weight, e.g. more than 95% of the THP

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and syntan. Where white leather is required, we prefer that the total tannages consist essentially of the THP and syntan. In particular we prefer the leather is not tanned with vegetable or mineral tannage.

The THP and polyhydroxy moderator may be used in conjunction with mineral tannages for example, a tannage containing a mixture of THP, chrome and moderator gives strongly synergistic tanning with high shrink temperatures and good area yield using only small amounts of chrome. Combinations of THP, moderator and aluminium salts are particularly preferred, especially aluminium triformate. The latter gives particularly high shrink temperatures. Alternatively the THP tannage may be used as a pretan or re-tan in a mineral tanning operation

The THP is preferably applied in the substantial absence of monomers or prepolymers capable of copolymerising with the THP such as phenol, urea, melamine or their precondensates with formaldehyde. For the purposes of this specification, "the substantial absence of monomers or prepolymers" means less than the minimum that would be capable of reacting or copolymerising with 50% of the THP, more preferably less than the minimum that would be required to react or copolymerise with 20%, e.g. less than 5% by weight based on the weight of THP, most preferably less than 2%, especially less than 1%.

The THP may be applied as a first tanning step with a syntan as a retan. Preferably the THP is applied to acidified skins following aqueous degreasing. E.g. the initial pH is typically below 5, e.g. below 4. We prefer that the pH be raised above 5 and preferably maintained above 6 for the main duration of the tanning process.

The skins are preferably agitated in the tanning liquor for a sufficient time to raise the shrink temperature above 75°C, more preferably above 80°C, most preferably above 85°C.

According to a particularly preferred embodiment the polyhydroxy compound is a surfactant such as a C₆₋₂₅ alkyl polyglycoside, sucrose ester or sorbitan ester and the

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THP and polyhydroxy compound are applied to undegreased or partially degreased skins, in order to effect degreasing or further degreasing thereof.

The tanned skins are typically washed with warm water and fat liquored using a suitable oil or blend of oils. Fat liquoring is normally carried out after dyeing.

The invention will be illustrated by the following examples.

Example 1

60g pickled bovine skin was rotated for 10 minutes with 15% based on the wet weight of skin of 8% sodium chloride solution. 3% based on the wet weight of skin of 75% by wt. THP solution and 1% sucrose was added and the mixture rotated for 3 hours. The solution was basified to pH 6.5 by adding 2.25% based on the wet weight of skin of sodium bicarbonate and the skins left to rotate overnight.

After the first 3 hours penetration was tested with sodium selenite solution. A pale, even colour indicated even penetration throughout the skin. In contrast a control using THPS without sugar gave only surface action.

After basification a further sample was tested with sodium selenite solution. A strong orange colouration evenly distributed was observed throughout the skin. THP alone gave an uneven colouration. The example was repeated three times using respectively maltose, lactose and D-glucose instead of sucrose, all with similar results.

All the samples gave similar shrink temperatures within the range 81-83°C except for the lactose which gave a shrink temperature of 78°C, but the THP control felt hard and boardy while the other samples were soft and pliable. All the examples of the invention 1 to 12 described herein showed a reduction in formaldehyde level of between 30 and 45% compared with the control.

Example 2

400g of brine pickled pelt was rotated for 10 minutes with 150% by weight of water based on the wet weight of the skins at pH 4.6. 3% THP based on the wet weight of skins was added as a 75% aqueous solution together with 3%, based on the wet weight of skins, of glycerol. After 3 hours rotation at 35°C the pH was 3.38 and the shrink temperature was 58°C. After basifying to pH 4.2 with sodium bicarbonate the shrink temperature was 64°C. Basification to pH 5.5 gave a shrink temperature of 68°C. Further basification to pH 6.5 gave a shrink temperature of 77°C.

Testing with sodium selenite showed even penetration and the product was soft and pliable.

Example 3

200gm bovine pickled pelt was rotated overnight with a mixture of 3% based on the wet weight of skin of 75% wt/wt aqueous THPS and 3% based on the wet weight of skin of a 30,000 to 70,000 mole wt polyvinyl alcohol at a pH of 3.5. The pH was then raised to 6.5 with sodium carbonate. After a further hour's rotation the product was full and soft with a shrink temperature of 79°C.

Example 4

Example 3 was repeated using a polyvinyl alcohol of molecular weight greater than 70,000. The product was soft but less full than that of Example 3. The shrink temperature was 75°C.

Example 5

Example 3 was repeated using a 6:1 molar THPS/urea condensate instead of THPS. The product was soft and full. The shrink temperature was 78°C.

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Example 6

Example 3 was repeated using a 3:1 molar THPS/urea condensate. The product was full but less soft than Example 3. The shrink temperature was 76°C.

Example 7

Example 6 was repeated using a PVA with a molecular weight greater than 70,000. The product was less tight and hard than that obtained using the urea/THP condensate alone, but not as full as that of Example 6. The shrink temperature was 75°C.

Example 8

Example 3 was repeated using polyvinylpyrrolidone of mole weight approximately 10,000 in place of PVA. The product was full and soft and had a shrink temperature of 77°C.

Example 9

Example 3 was repeated using a 2:1 molar THPS/urea condensate instead of THPS. The product was soft and full, with little residual odour.

Example 10

Example 3 was repeated using dialdehyde starch in place of PVA. The product was full and soft.

Example 11

60gm pickled bovine hide was rotated for 10 mins with 150% float containing 8% sodium chloride. 3% THPS/urea condensate (6:1 mole ratio) and 1% dextrose were

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added and the rotation continued for 3 hours. Small additions of sodium carbonate were made until the pH was 6.5.

Selenium indicator showed good penetration at the end of the three hour rotation. The shrink temperature was 75°C.

The leather was split and shaved prior to retanning with an acrylic resin sytan. The leather gave a more even split than a control without the moderator, and required less shaving.

The retanned leather was washed and fat liquored using 10% by weight of wet skin of a mixed sulphated/sulphited oil fat liquor, and finally dried.

The leather was a very full and soft white leather.

Example 12

100gm bovine skins and 150% water based on wet weight of skin were rotated for 10 minutes with 8% sodium chloride based on wet weight of skin.

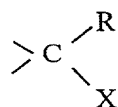
3% (based on the wet weight of skin) of a 76% aqueous THPS solution and 3% of sodium hexametaphosphate were rotated for 3 hours at pH 2 to 4 and then basified to pH 6.5.

The product was full and soft with a shrink temperature of 78°C.

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CLAIMS

- 1) A mixture of THP or of a THP salt, THP condensate or THP analogue with at least one water soluble moderator selected from (A) metaphosphate salts and (B) aliphatic polyhydroxy or polypyrrolidone compounds comprising at least three adjacent, geminal or vicinal



groups where R is hydrogen or an alkyl group, and X is a

hydroxyl or pyrrolidone group.

- 2) A mixture according to claim 1 comprising a condensate of from 2 to 6 moles THP with 1 of urea.
- 3) A mixture according to either of claims 1 or 2 comprising a moderator selected from sorbitol, glycerol, trimethylolpropane, pentaerythritol, mannitol, mono and di- saccharide sugars, dialdehyde starch, alginates, polyvinyl alcohol and polyvinyl pyrrolidone.
- 4) A method of tanning leather which comprises contacting collagen with a tannage comprising a mixture according to any foregoing claim.
- 5) A method according to claim 4 wherein said mixture is used prior to, simultaneously with, or after tanning with a syntan and/or a mineral tannage.
- 6) A method according to either of claims 4 and 5 wherein the leather is subsequently fat liquored.
- 7) A method according to anyone of claim 4 to 6 wherein the leather is subsequently shaved and/or split.

APPLICATION FOR UNITED STATES LETTERS PATENT
POST-FILED PCT Declaration and Power of Attorney (35 U.S.C. 371(c)(4))
PCT Application - United States Designated Office

As a below named inventor, I declare that:

My residence, post office address and citizenship are as stated below next to my name; I believe that I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled:
"LEATHER TANNING"

described and claimed in International Application number PCT/EP00/05471 filed June 14, 2000
and, if it was amended, as amended on

I have reviewed and understand the contents of said specification, including claims.

I acknowledge the duty to disclose information which is material to patentability as defined in 37 CFR §1.56.

I claim priority benefits under 35 USC §119 of: (i) any foreign application(s) for patent or inventor's certificate listed below; or (ii) any United States provisional application(s) listed below; and have also identified below any foreign application(s) for patent or inventor's certificate, or PCT international application having a filing date before that of the application(s) on which priority is claimed.

COUNTRY	APPLICATION NUMBER	DATE (day, month, year)	PRIORITY CLAIMED
Great Britain	9914139.2	18 June 1999	yes <input checked="" type="checkbox"/> no <input type="checkbox"/>
Great Britain	9918243.8	04 August 1999	yes <input checked="" type="checkbox"/> no <input type="checkbox"/>
Great Britain	9918698.3	10 August 1999	yes <input checked="" type="checkbox"/> no <input type="checkbox"/>

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

I appoint the following attorneys to prosecute this application and to transact all business in the U.S. Patent & Trademark Office connected therewith: Leonard Holtz, Reg. No. 22,974; Herbert Goodman, Reg. No. 17,081; Thomas Langer, Reg. No. 27,264; Marshall J. Chick, Reg. No. 26,853; Richard S. Barth, Reg. No. 28,180; Douglas Holtz, Reg. No. 33,902; and Robert P. Michal, Reg. No. 35,614.

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